Synthesis of Acerogenin C and (+)-Acerogenin A, Two Macrocyclic Diarylheptanoid Constituents of *Acer nikoense*

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The macrocyclic ketone acerogenin C (3) and the corresponding alcohol (+)-accerogenin A (1), diarylheptanoid constituents of the maple *Acer nikoense* were synthesized. The key

steps were the selective reduction of the double bond of an α,β -unsaturated ketone (10) and macrocyclization of an iodophenol (13) by a modified Ullmann diarylether synthesis.

Introduction

Studies continued by now over two decades by Kubo, Nagai, and their associates on the elucidation of the constituents of *Acer nikoense* MAXIM (Aceraceae), a maple indigenous in Japan, led to the isolation of several macrocyclic diarylheptanoids, namely the acerogenins A (1), B (2), C (3), D (4), and E (6) and seven of their glycosides, the acerosides.^[1]

Scheme 1

1 R¹ = H, R² = R³ = H, R⁴ = OH (acerogenin A) 2 R¹ = H, R² = OH, R³,R⁴ = H (acerogenin B) 3 R¹ = H, R² = H, R³,R⁴ = O (acerogenin C)

4 R¹ = H, R² = OH, R³,R⁴ = O (acerogenin D) 5 R¹ = Me, R² = H, R³,R⁴ = O

The relatively simple structure of the acerogenins and the fact that at the time of starting our work no synthesis of a natural macrocyclic diarylheptanoid with a diarylether linkage had been reported, prompted us to synthesise these compounds. In the meantime we have completed the syntheses of three other macrocyclic diarylheptanoids of more

complex structure (garugamblin-1^[2], garugamblin-2^[3], and garuganin III^[4]). Attempts at the synthesis of acerogenins were reported in a separate paper.^[5] Now we report the syntheses of acerogenin C (3) and (+)-acerogenin A (1).

Two basic strategies can be envisaged for the synthesis of acerogenins A to C: (i) Linking the two aryl rings first, preferably between C-1 and O-2 (bond a) followed by elaboration of the aliphatic chain and ring closure to form the C-8–C-9 bond (b) in the case of acerogenin B or the C-13–C-14 bond (c) with acerogenin A and C. (ii) Preparation of a suitably substituted 1,7-diarylheptane derivative and formation of the diarylether bond at a late stage of the synthesis. Failure of strategy (i)^[5] prompted us to turn to strategy (ii). This approach, combined with the use of a recently introduced reagent [Cu^IH·Ph₃P hexamer)]^[6] for the selective reduction of α , β -unsaturated ketones to saturated ketones enabled finally the synthesis of acerogenin C (3) and (+)-acerogenin A (1).

Results and Discussion

First the hydroxyester **7** was prepared in three steps from 3-benzyloxy-4-methoxybenzaldehyde and 4-triphenylphosphoniobutanoic acid. ^[5] The hydroxyl group of **7** was blocked by tetrahydropyranylation and the product **8** reacted with diethyl methylphosphonate to give the ketophosphonate **9**. Condensation of the latter with 4-iodobenzaldehyde ^[7] gave the protected enone **10**, or using acidic workup, the hydroxyenone **11**. Reduction of the latter with either diisobutylaluminium hydride or LiAlH₄ at low temperature gave only the corresponding allylic alcohol (**14**). Treatment, however, of **10** with Cu(I)H.Ph₃P hexamer ^[6] in benzene saturated with water provided the desired heptanone **12**. Removal of the tetrahydropyranyl group from **12** gave our key intermediate the diarylheptanone **13**.

Macrocyclization was carried out by a modification of the classical Ullmann reaction, as recommended by Boger Scheme 2

et al. [8] i.e. using CuBr·Me₂S complex as catalyst. This gave finally the cyclized product 5 in low yield. M.p. and ¹H-NMR spectrum of 5 was in complete agreement with data for acerogenin C methyl ether. [1b]

Methyl ether **5** was demethylated by heating with pyridine hydrochloride at 210 °C to give the phenol **3**, that proved to be identical (m.p., ¹H NMR) with accrogenin C.^[1d]

Reduction of acerogenin C (3) with NaBH₄ led to racemic acerogenin A (1) of m.p. 186–188°C (ref. [1b] for (+)-1 151–152°C). M.p. of (+)-1 indicated that it formed a racemic phase. Note that the m.p. of acerogenin B (2), a racemic compound is very similar (179°C). [1c]

It was already observed by Nagai et al. [1b] that in the ¹³C-NMR spectrum of acerogenin A (1) unsubstituted carbons of the *para* disubstituted benzene ring gave rise to four signals. Now we observed the same phenomenon for the corresponding proton signals too. Inspection of a model indicated clearly that rotation of this benzene ring is severely hindered. Although the ground state conformation of acerogenin C (3) is also chiral, no such signal splitting was observed, probably owing to the flexibility of the seven carbon chain

We have been informed that Dr. Jieping Zhu and his coworkers (Gif-sur-Yvette, France) have also accomplished very recently a total synthesis of acerogenin C.

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Experimental Section

General: NMR: Bruker DRX 500 (500.15 and 125.77 MHz for ¹H and ¹³C NMR, respectively). Signal assignments of **1** were based on COSY and HMQC spectra. M.p.s were not corrected, evaporations were carried out in vacuum, for chromatography silica gel 60 (Merck) was used.

Methyl 5-[4-Methoxy-3-(tetrahydropyran-2-yloxy)phenyl]pentanoate (8): Hydroxyester 7⁵ (4.4 g, 17.6 mmol), pyridinium p-toluenesulphonate (80 mg) and 3,4-dihydropyran (4.4 ml) were left standing for 48 h. After neutralization with triethylamine the solu-

tion was evaporated and the residue purified by chromatography (eluent hexane-Me₂CO 3:1) to give the ether **8** as an oil (3.9 g, 66%). - ¹H NMR (CDCl₃): δ = 1.60 (mc, 4 H, 3,4-CH₂), 1.8–2.2 (m, 6 H, 3'',4'',5''-CH₂), 2.30 (t, J = 7.3 Hz, 2 H, 5-CH₂), 2.55 (t, J = 7.3 Hz, 2 H, 2-CH₂), 3.55 (m, 1 H) and 4.00 (m, 1 H, 6''-CH₂), 3.66 (s, 3 H, CO₂CH₃), 3.83 (s, 3 H, OCH₃), 5.55 (m, 1 H, 2''-H), 6.6–6.75 (m, 2 H, 5',6'-H), 6.94 (br. s, 1 H, 2'-H). -C₁₈H₂₆O₅ (322.4): calc. C 67.06, H 8.13; found C 67.23, H 7.98.

6-[4-Methoxy-3-(tetrahydropyran-2-yloxy)phenyl]-2oxohexylphosphonate (9): a) To a solution of methyl diethylphosphonate (2.6 ml, 17.4 mmol) in dry THF first BuLi (10.6 ml of a 1.6 M solution in hexane, 17 mmol) was added at −70°C, followed by a solution of ester 8 (2.9 g, 8.7 mmol) in THF (10 ml). The mixture was allowed to warm up to room temperature, evaporated, the residue dissolved in CH₂Cl₂ and the solution washed with aq. NaHCO₃ solution. Evaporation gave the phosphonate 9 (3.6 g, 94%) as a viscous oil. – ¹H NMR (CDCl₃): $\delta = 1.33$ (t, J = 7.1Hz, 6 H, CMe), 1.59 (m, 4 H, 4,5-CH₂), 1.5-1.9 (m, 6 H, 3'',4'',5''-CH₂), 2.52 (t, J = 6.7 Hz, 2 H, 3-CH₂), 2.65 (t, J = 6.7Hz, 2 H, 6-CH₂), 3.05 (d, J = 22.7 Hz, 2 H, PCH₂), 3.50 (m, 1 H) and 4.00 (m, 1 H, 6"-CH₂), 3.85 (s, 3 H, OCH₃), 4.13 (mc, 4 H, CH_2Me), 5.63 (m, 1 H, 2"-H), 6.63 (dd, J = 8.1 and 1.5 Hz, 1 H, 6'-H), 6.75-6.80 (m, 2 H, 2',5'-H). $-C_{22}H_{35}O_7P$ (442.3): calcd. C 59.72, H 7.97; found C 59.57, H 7.78.

b) Diethyl 6-(3-hydroxy-4-methoxyphenyl)-2-oxohexylphosphonate [5] (6.1 g, 17 mmol), was converted to the tetrahydropyranyl ether **9** as described for **8**.

(E)-1-(4-Iodophenyl)-7-[4-methoxy-3-(tetrahydropyran-2yloxy)phenyl]hept-1-en-3-one (10): To a solution of phosphonate 9 (4.5 g, 8.7 mmol) and 4-iodobenzaldehyde (2.0 g, 8.6 mmol) in MeOH (90 ml) 1 M NaOMe (8.7 ml) was added. After 30 min the solution was evaporated, the residue treated with water and taken up in CH₂Cl₂. Evaporation, chromatography (eluent hexane/ Me₂CO, 3:1) and trituration with diethyl ether gave enone 10 (2.0 g, 44%) as colorless crystals of m.p. 74-75°C. -1H NMR (CDCl₃): $\delta = 1.6-1.8$ (m, 6 H, 3''', 4''', 5'''-CH₂), 1.7 (mc, 4 H, 5,6-CH₂), 2.59 (t, J = 7.2 Hz, 2 H,) and 2.68 (t, J = 6.7 Hz, 2 H, 4,7-CH₂), 3.61 (m, 1 H) and 4.04 (m, 1 H, 6""-CH₂), 3.85 (s, 3 H, Me), 5.40 (m, 1 H, $2^{\prime\prime\prime}$ -H), 6.73 (d, J = 16.0 Hz, 1 H, 2-H), 6.79-6.83 (m, 2 H, 5",6"-H), 7.27-7.29 (m, 3 H, 3',5', 2"-H), 7.45 (d, J = 16.0 Hz, 1 H, 1-H), 7.75 (d, J = 7.8 Hz, 2-H, 2',6'-H). - C₂₅H₂₉IO₄ (520.4): calcd. C 57.70, H 5.62; found C 57.84, H 5.70.

7-[4-Methoxy-3-(tetrahydropyran-2-yloxy)phenyl]-1-(4-iodophenyl)heptan-3-one (11): A solution of heptenone 10 (2.0 g, 3.85 mmol) in benzene (40 ml) to which water (1.0 ml) was added was boiled in a stream of argon for 15 min. The mixture was cooled to r.t., [Cu^IH·Ph₃P]₆ (2.3 g, 1.17 molar equivalents) added and stirring continued for 24 h. The mixture was filtered through a layer of Cellite and evaporated (1.6 g, 80%). A small sample was purified for analysis and spectroscopy. Viscous oil. - ¹H NMR (CDCl₃): $\delta = 1.57$ (mc, 4 H, 5,6-CH₂), 1.5–1.9 (m, 6 H, 3''',4''',5'''-CH₂), 2.37 (t, J = 7.2 Hz, 2 H), and 2.50 (t, J = 7.3 Hz, 2 H, 4,7-CH₂), 2.67 (t, J = 7.3 Hz, 2 H,) and 2.81 (t, J = 7.4 Hz, 2 H, 1,2-CH₂), 3.53 (m, 1 H) and 4.00 (m, 1 H, 6"'-CH2), 3.86 (s, 3 H, OMe), 4.95 (m, 1 H, 2'''-H), 6.61 (dd, J = 8.1, 2.1 Hz, 1 H, 6''-H), 6.73(d, J = 2.1 Hz, 1 H, 2''-H), 6.75 (d, J = 8.1 Hz, 1 H, 5"-H), 6.92(d, J = 8.2 Hz, 2 H, 2', 6' - H), and 7.5 (d, J = 8.2 Hz, 2 H, 3', 5' -H). - C₂₅H₃₁IO₄ (522.4): calcd. C 57.48, H 5.98; found C 57.25, H 5.64.

7-(3-Hydroxy-4-methoxyphenyl]-1-(4-iodophenyl)heptan-3-one (13): Unpurified heptan-one 12 (2.0 g) (containing Ph₃P arising

from the reducing agent) was boiled for 40 min in a mixture of MeOH (100 ml) and 10% HCl (10 ml). After 16 h the solution was decanted from the deposited crystals of Ph₃P, evaporated and the residue chromatographed (eluent hexane/Me₂CO, 3:1) to give the iodophenol **13** (1.0 g, 59% calculated on **9**) as colourless crystals of m.p. 176–178°C. $^{-1}$ H NMR (CDCl₃): δ = 1.56 (mc, 4 H, 5,6-CH₂), 2.37 (t, J = 7.2 Hz, 2 H) and 2.50 (t, J = 7.3 Hz, 2 H, 4,7-CH₂), 2.69 (t, J = 7.3 Hz, 2 H) and 2.81 (t, J = 7.3 Hz, 2 H, 1,2-CH₂), 3.86 (s, 3 H, OMe), 6.62 (dd, J = 8.1, 2.1 Hz, 1 H, 6"-H), 6.74 (d, J = 2.1 Hz, 1 H, 2''-H), 6.76 (d, J = 8.1 Hz, 1 H, 5''-H), 6.93 (d, J = 8.3 Hz, 2 H, 2',6'-H), and 7.59 (d, J = 8.3 Hz, 2 H, 3',5'-H). - C₂₀H₂₃IO₃ (438.3): calcd. C 54.81, H 5.29; found C 54.60, H 5.35.).

4-Methoxy-2-oxatricyclo[13.2.2.1^{3,7}]eicosa-3,5,7(20),15,17,18hexaen-12-one, Acerogenin C Methyl Ether (5): To a solution of iodophenol 13 (500 mg, 1.14 mmol) in dry benzene (10 ml) KOtBu (180 mg, 1.6 mmol) was added and the solution evaporated to dryness. Me₂S·CuBr (2.35 g, 11.4 mmol) was added and dry pyridine (250 ml) filtered through a column of powdered molecular sieve. The mixture was stirred under argon for 8 h in a bath of 130°C, the solvent evaporated, the residue treated with 10% HCl and extracted with CH₂Cl₂. Chromatography (eluent hexane/acetone, 3:1) gave methyl ether 5 (56 mg, 16%) as colourless needles of m.p. 122-124°C (ref.^[1b] m.p. 124°C). - ¹H NMR (CDCl₃): δ = 1.11 $(q, J = 7.4 \text{ Hz}, 2 \text{ H}, 9-\text{CH}_2)$ and 1.37 (mc, 2 H, 10-CH₂), 1.90 (t, $J = 8.1 \text{ Hz}, 2 \text{ H}, 11\text{-CH}_2$, 2.44 (t, $J = 5.6 \text{ Hz}, 2 \text{ H}, 8\text{-CH}_2$), 2.60 $(t, J = 6.6 \text{ Hz}, 2 \text{ H}, 13\text{-CH}_2), 2.98 (t, J = 6.3 \text{ Hz}, 2 \text{ H}, 14\text{-CH}_2),$ 3.94 (s, 3 H, OMe), 5.64 (d, J = 1.6 Hz, 1 H, 20-H), 6.64 (dd, J =8.1, 1.6 Hz, 1 H, 6-H), 6.81 (d, J = 8.1 Hz, 1 H, 7-H), 7.02 (d, J =8.4 Hz, 2 H), and 7.17 (d, J = 8.4 Hz, 2 H, 17,18- and 16,19-H); $(ref.^{[1b]} \, ^1H \, NMR \, (CDCl_3): \delta = 1.09, 1.37, 1.90, 2.44, 2.61, 2.96,$ 3.93, 5.65, 6.63, 6.82, 7.01, 7.20). $- {}^{13}$ C NMR (CDCl₃): $\delta = 20.4$ (C-10), 27.5 (C-9), 31.3 (C-8), 32.3 (C-14), 44.6 (C-13), 46.3 (C-11), 56.3 (Me), 112.4 (C-5), 117 4 (C-20), 122.0 (C-6), 123.7 (C-17,18), 130.6 (C-16,19), 133.7 (C-7), 137.1 (C-15), 146.8 (C-4), 150.9 (C-3), 156.9 (C-1), 212.1 (CO) (ref. [1c] 20.5, 27.6, 31.5, 32.3, 44.6, 46.2, 56.4, 112.7, 117.7, 122.2, 123.9, 130.8, 133.9, 137.3, 147.1, 151.2, 157.4, 212.0). - C₂₀H₂₂O₃ (310.4): calcd. C 77.39, H 7.14; found C 77.52, H 7.23.

12-Oxo-2-oxatricyclo[13.2.2.1^{3,7}]eicosa-3,5,7(20),15,17,18hexaen-4-ol, Acerogenin C (3): A mixture of methoxyketone 5 (50 mg) and pyridine · HCl (200 mg) was heated under argon at 210 °C for 3 h. After adding water (4 ml) the raw product was extracted with CH₂Cl₂ and purified by TLC (eluent benzene/EtOAc, 20:1) to give 3 (28 mg, 59%), as colourless crystals of m.p. 114-116°C, (ref. [1d] m.p. 116°C). - ¹H NMR (CDCl₃): $\delta = 1.05$ (q, J = 7.4Hz, 2 H, 9-CH₂), and 1.37 (mc, 2 H, 10-CH₂), 1.90 (t, J = 8.1 Hz, 2 H, 11-CH₂), 2.45 (t, J = 5.6 Hz, 2 H, 8-CH₂), 2.61 (t, J = 6.6, Hz 2 H, 13-CH₂), 3.00 (t, J = 6.3 Hz, 2 H, 14-CH₂), 5.64 (d, J =1.6 Hz, 1 H, 20-H), 6.62 (dd, J = 8.2, 1.8 Hz, 1 H, 6-H), 6.85 (d,J = 8.2 Hz, 1 H, 7-H, 7.00 (d, J = 8.3 Hz, 2 H), and 7.18 (d, J = 8.2 Hz, 1 H)8.3 Hz, 2 H, 17,18- and 16,19-H); (ref. [1d] 1 H NMR (CDCl₃): δ = 1.09, 1.37, 1.90, 2.44, 2.61, 2.96, 5.55, 5.64, 6.63, 6.85, 6.99, 7.19.). $- {}^{13}$ C NMR (CDCl₃): $\delta = 20.4$ (C-10), 27.4 (C-9), 31.5 (C-8), 32.2 (C-14), 44.5 (C-13), 46.3 (C-11), 115.5 (C-5), 117 0 (C-20), 122.8 (C-6), 123.5 (C-17,18), 130.7 (C-16,19), 133.0 (C-7), 137.6 (C-15), 143.0 (C-4), 149.1 (C-3), 156.8 (C-1), 212.8 (CO). $-C_{19}H_{20}O_3$ (296.4): calcd. C 77.00, H 6.80; found C 69.85, H 6.88.

rac-2-Oxatricyclo [13.2.2. $I^{3.7}$]eicosa-3,5,7(20),15,17,18-hexaen-4,12-diol, Acerogenin A (1): A solution of acerogenin C (3) (13 mg, 0.044 mmol) in EtOH (4 ml) was treated under argon with NaBH₄ (5 mg) for 1 h. Acidification with AcOH, evaporation, and TLC

(eluent benzene/EtOAc, 8:1) gave the alcohol 1 [(+)-acerogenin A] $(10 \text{ mg}, 71\%), \text{ m.p. } 186-188^{\circ}\text{C}; (\text{ref.}^{[1b]} \text{ for } (+)-1 \ 151-152^{\circ}\text{C}). -$ ¹H NMR (CDCl₃): $\delta = 0.838$ (m, 1 H) and 1.036 (m, 1 H, 10-CH₂), 0.949 (m, 1 H) and 1.095 (m, 1 H, 11-CH₂), 1.26 (m, 1 H) and 1.62 (m, 1 H, 9-CH₂), 1.62 (m, 1 H) and 1.88 (dq, J 14.1, 4.4 Hz, 1 H, 13-CH₂), 2.46 (m, 2 H, 8-CH₂), 2.69 (dt, J 12.8, 4.1 Hz, 1 H), and 2.99 (dt, J = 13.2, 3.9 Hz, 1 H, 14-CH₂), 3.35 (q, J =5.5 Hz, 1 H, 12-H), 5.63 (br. s, 1 H, 20-H), 6.60 (d, J = 8.2 Hz, 1 H, 6-H), 6.83 (d, J = 8.1 Hz, 1 H, 5-H), 6.94 (dd, J = 8.8, 3.1 Hz, 1 H) and 7.21 (dd, J = 8.2, 2.5 Hz, 1 H, 16,19- or 17,18-H), 7.16 (dd, J = 8.8, 3.1 Hz, 1 H) and 7.29 (dd, J = 8.2, 2.5 Hz, 1 H)16,19- or 17,18-H); [(ref. [1d] (in C₅D₅N at 100 MHz): 0.68-1.12, 1.12-1.48, 1.48-1.84, 1.84-2.24, 2.24-2.52, 2.52-2.99, 3.24-3.56, 5.84, 6.64, 6.93, 7.06-7.22]. - ¹³C NMR (CDCl₃): $\delta =$ 24.9 (C-10), 27.9 (C-9), 31.7 (C-8), 32.5 (C-14), 39.5 (C-11), 40.5 (C-13), 71.1 (C-12), 115.1 (C-5), 115.3 (C-20), 122.5 (C-6), 123.1 and 124.1 (C-17,18), 130.1 and 131.6 (C-16,19), 133.9 (C-7), 139.6 (C-15), 142.9 (C-4), 149.0 (C-3), 155.7 (C-1), [ref. [1c] (in C_5D_5N at 25 MHz) $\delta = 25.3$, 28.5, 32.0, 32.7, 39.7, 40.9, 69.8, 116.7, 117.1, 122.5, 123.0, 124.2, 130.2, 131.8, 132.8, 139.7, 145.1, 150.7, 156.6]. - C₁₉H₂₂O₃ (298.4): calcd. C 76.48, H 7.43; found C 76.58, H 7.50.

(*E*)-7-(3-Hydroxy-4-methoxyphenyl)-1-(4-iodophenyl)hept-1-en-3-one (13): This compound was prepared as 10, except that the crude product was hydrolyzed with a 5% citric acid solution. Colourless crystals, m.p. 98–100°C. – ¹H NMR (CDCl₃): δ = 1.6–1.8 (m, 4 H, 5,6-CH₂), 2.56 (t, J=7.2 Hz, 2 H) and 2.65 (t, J=6.7 Hz, 2 H, 4,7-CH₂), 3.83 (s, 3 H, Me), 6.71 (d, J=16.1 Hz, 1 H, 2-H), 6.78 (dd, J=8.1, 1.5 Hz, 1 H, 6''-H), 6.81 (d, 8.1 Hz, 1 H, 5''-H), 6.96 (d, J=1.5 Hz, 1 H, 2''-H), 7.26 (d, J=8.3 Hz, 2 H, 3',5'-H), 7.44 (d, J=16.1 Hz, 1 H, 1-H), 7.75 (d, J=8.3 Hz, 2 H, 2',6'-H). – C₂₀H₂₁IO₃ (436.3): calcd. C 55.06, H, 4.85; found C 54.91, H, 4.66.

(*E*)-7-(3-Hydroxy-4-methoxyphenyl)-1-(4-iodophenyl)hept-1-en-3-ol (14): To a solution of 11 (120 mg, 0.23 mmol) in dry THF (5 ml) iBu_2AlH (0.46 mmol, 0.65 ml of a 0.7 m solution in toluene) was added at $-70\,^{\circ}$ C. After 1 h the reaction mixture was quenched with MeOH and then with 5% HCl. Evaporation, extraction with CH₂Cl₂ and chromatography of the crude product (eluent benzene/EtOAc, 20:1) gave the alcohol 14 (100 mg, 83%) as a resin. Reduction of 11 with LiAlH₄ at r.t. produced the same compound. $-^{1}$ H NMR (CDCl₃): δ = 1.5-1.65 (m, 4 H, 5,6-CH₂), 2.15 (m, 2 H, 4-CH₂), 2.51 (t, J = 6.7 Hz, 2 H, 7-CH₂), 3.69 (m, 1 H, 3-H), 3.85 (s, 3 H, OMe), 6.19 (dd, J = 15.9, 6.6 Hz, 1 H, 2-H), 6.55 (d, J = 15.9 Hz, 1 H, 1-H), 6.62 (dd, J = 8.1, 1.6 Hz, 1 H, 6''-H), 6.73 (d, J = 8.1 Hz, 1 H, 5''-H), 6.74 (d, J = 1.6 Hz, 1 H, 2''-H), 7.12 (d, J = 8.3 Hz, 2 H, 3',5'-H), 7.65 (d, J = 8.3, 2 H, 2',6'-H). $- C_{20}H_{23}IO_3$ (438.3): calcd. C 54.81, H 5.29; found C 54.90, H 5.09.

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